

WATER FOR USE WITH CEMENT

1. SCOPE: This test method outlines the procedure for sampling and testing water for use with cement.

2. APPARATUS AND MATERIALS:

- 2.1. 0.1 Normal Sulfuric Acid
- 2.2. 0.02 Normal Sodium Hydroxide
- 2.3. Methyl Orange Indicator
- 2.4. Phenolphthalein Indicator
- 2.5. Potassium Chromate Indicator

- 2.6. 0.1 AgNO₃

- 2.7 pH Meter: Calibrated at pH 4.0, pH 7.0, and pH 10.0

3. SAMPLE: Water samples are received in a 0.95 liter (quart) glass or plastic containers.

4. PROCEDURE:

- 4.1. Organic and Inorganic Solids: - (Run in Duplicates)

- 4.1.1. Heat a ~~No. 150 ml~~ porcelain crucible in a ~~bright red~~ muffle furnace at 950 °C for 15 minutes. Cool in desiccator, and carefully weigh to 0.0001 grams. Shake sample well, and measure 25 ml in graduated cylinder. Immediately transfer to weighed crucible. Place crucible on steam bath, or on asbestos pad on hot plate, and evaporate to dryness. Place in oven at temperature of ~~100—100~~ 110 °C for one hour. Cool in desiccator and re-weigh.

Gain in Weight x 4 = Total Solids

- 4.1.2. Heat crucible and total solids in ~~bright red~~ muffle furnace at 950 °C for 30 minutes. Cool in desiccator and re-weigh.

Loss in weight x 4 = Organic Solids

- 4.1.3. Inorganic solids are determined by subtracting organic solids from total solids.

Organic solids and inorganic solids are reported separately.

4.2. Acidity or Alkalinity:

4.2.1. Most water samples received are alkaline. Test results are reported as percent total alkalinity in terms of calcium carbonate.

4.2.2. Measure 100 ml of sample in a graduated cylinder after shaking well. Transfer to porcelain casserole, and add 3 drops of methyl orange indicator. Titrate to orange end point by adding 0.1N sulfuric acid drop by drop from a burette.

$$\text{ml } 1/10 \text{ N H}_2\text{SO}_4 \times .005005 = \text{Percent Alkalinity as CaCO}_3$$

4.2.3. If acidity is to be determined, the same general procedure as above may be used by titrating with 0.02 N sodium hydroxide solution using phenolphthalein as the indicator.

4.3. Chloride ion in water: Procedure in ASTM D 512 Method B Silver Nitrate.

4.4. pH of water

4.4.1. Place 100 ml of sample in 150 ml beaker.

4.4.2. Add small stir bar.

4.4.3. Place on stir plate and adjust stirrer to medium speed.

4.4.4. Rinse pH probe with D.I. water and place in beaker containing sample.

4.4.5. Allow reading to stabilize, record pH and temperature in °C.

5. CALCULATIONS: Calculations are given in the Procedures.

6. PRECAUTIONS: Any precautions necessary are found in the Procedures.

7. REPORT:

7.1. % Total Solids

7.2. % Organic Solids

7.3. % Inorganic Solids

7.4. % Alkalinity or Acidity as CaCO_3

7.5. % Chloride ion in water

7.6 pH and temperature

APPROVED

DIRECTOR
DIVISION OF MATERIALS

DATE 03/07/08

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Director

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